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# Asymmetric Synthesis of (+) and (-) trans-2,6-dimethylpiperidines.

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Abstract: (+) and (-) trans-2,6-dimethylpiperidines 9 have been prepared from diastereomeric lactams 6 and 11, both available from (R)-(-)-phenylglycinol 4. Diastereoselective C-2 alkylation afforded oxazolidines 7 and 12. Reduction of 7 with LiAlH4 or NaBH4 led to trans-2,6-8 with total retention of configuration. When 12 was reduced with NaBH4, trans derivative 13a was isolated as the major isomer. A mechanism is proposed to explain the different results. © 1997 Elsevier Science Ltd.

#### INTRODUCTION

Recently we described a method for the preparation of optically pure 2,3-disubstituted piperidines 3, involving C-2 functionalization of chiral, non-racemic alkylated lactams 1.1 The intermediate oxazolidines 2 were obtained by addition of a Grignard reagent, followed by cyclization of the hydroxyl group onto the transient carbinolamine (Scheme 1). The diastereoselectivity observed during the formation of 2 was dependent on the nature of the organometallic species and the C-3 substituent. When LiAlH4 was used, opening of the oxazolidine ring was diastereoselective and occurred with retention of configuration, as previously observed by Meyers.<sup>2</sup>

Scheme 1

At the same time, we developed two new methods for the preparation of (R)- and (S)-6-methyl lactams 6 and 11, starting from (R)-(-)-phenylglycinol 4 (Scheme 2). One methodology<sup>3</sup> involved the formation of oxazolopiperidine  $5^{2a}$ , followed by a diastereoselective reduction. The other one<sup>4</sup> was based on the diastereoselective alkylation-reduction sequence of cyano-oxazolopiperidine 10 (CN(R,S)) method<sup>5</sup>) leading to an oxazolidine intermediate which was submitted to an oxidation process (Br<sub>2</sub>, HO<sup>-</sup>) to furnish lactam 11.

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Scheme 2

At this point of our results, we considered applying the carbonyl functionalization method to 6 and 11 in order to get 2,6-disubstituted piperidines (Figure 1). It was particularly interesting to study the influence of the C-6 configuration on the diastereoselectivity of the alkylation and reduction steps. If the *cis-trans* ratio was observed to be different in the two series, one can expect to develop an alternative method to the preparation of *cis* and *trans* 2,6-disubstituted piperidines. The first results in the 2,6-dimethylpiperidine series are reported in this paper.

Figure 1

In the literature, numerous racemic approaches are described. Most of them consist in reducing 2,6-dimethylpyridine to lead to a mixture of *trans* and *cis* 2,6-dimethylpiperidines in 70-85 % yield<sup>6</sup>. The diastereomeric excess depends on the reducing agents ( $H_2$  in the presence of Ni-Al, Ru/C, Na/EtOH, Zn/AcOH/Ac<sub>2</sub>O). Another strategy is based on reductive amination of a 2,6-diketone<sup>7</sup>. The major product is the *cis* isomer (70% yield). Few asymmetric syntheses were developed<sup>8,9</sup>. The most efficient consists in synthesizing a diol from an asymmetric sulfone<sup>9b</sup>. After tosylation, the product was cyclized with benzylamine and hydrogenolyzed to lead to *trans*-2,6-dimethylpiperidine in 89 % yield and with a high level of enantioselectivity. Recently, a short synthesis of *trans*-(2*S*, 6*S*)-lupetidine was achieved in four steps from a chiral pyridinium salt derived from (R)-(-)-phenylglycinol. A *trans*-2,6-dimethyl- $\Delta$ <sup>3</sup>-piperideine was obtained in 50% d.e. after successive alkylations of the pyridinium salt and of the resultant tetrahydrooxazolopyridine<sup>8</sup>.

## RESULTS AND DISCUSSION

Oxazolidine 7 was obtained by cyclisation of the alcoholate of 6 in the presence of MeMgCl (7 eq., 3M/THF) in THF in 55% yield (Scheme 3). Only one diastereomer was detected by NMR, (d.e. superior to 98%). The stereochemistry of this compound was determined by nOe experiments; irradiation of C-8a methyl induced a nOe on H-5 and aromatic protons. Reduction of oxazolidine 7 with LiAlH4 or NaBH4 led to alcohol 8 in 82 % yield and a diastereomeric excess superior to 98 %. Hydrogenolysis (H<sub>2</sub>, Pd/C 10 %) of 8 afforded (-) 9 in 60 % yield. The  $[\alpha]_D$  value is in agreement with the previously reported value  $^{10}$ .

The same strategy was applied to lactam 11. We first observed that this compound was less reactive towards MeMgCl than diastereomeric compound 6. When the reaction was performed in THF, a equimolar mixture of 12a and 12b was obtained in poor yield; in Et<sub>2</sub>O the alcoholate was not soluble and the reaction was very slow. Finally, we prepared the alcoholate in THF (NaH, 1.2 eq.), then added MeMgI (10 eq.) in Et<sub>2</sub>O at rt. Under these conditions, the oxazolidines 12a and 12b were obtained in a 20/80 ratio in 45 % yield (70 % from reacting lactam).

Scheme 3

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To explain the diastereoselectivity observed during formation of oxazolidines 7 and 12, it is necessary to consider the intermediate iminium ions 15 and 16 (Scheme 4). They have to adopt one of the two conformations  $\bf A$  or  $\bf B$  in order to cyclize to oxazolidines. Strong interactions between phenyl ring and C-6 substituents disfavour conformation  $\bf A$ . Cyclization can then occur via intermediate  $\bf B$  by a stereoelectronically controlled attack of the nucleophile leading to compounds 7 and 12 $\bf b$ . The poorest diastereoselectivity observed for 12 (60 %) is the result of an increased allylic interaction between chiral auxiliary and C-6 methyl which adopted a pseudo equatorial position (16 $\bf B$ ,  $\bf R' = \bf Me$ ).

Scheme 4

As the two diastereomers 12a and 12b turned out to be difficult to separate, the reduction step was conducted on the mixture of oxazolidines. When LiAlH4 was used, the disubstituted piperidines 13a and 13b were obtained in 68 % yield with the same 20:80 ratio (Table).

Table: Reduction of oxazolidines 12a,b.

Reducing agent	Yield (%)	13a	13b
LiAlH4 / THF	68	20	80
NaBH4 / MeQH	72	85	15

This result, reflecting the stereochemical homogenity of the oxazolidines, is in agreement with Meyers' report<sup>2</sup>, and is a consequence of the virtually complete retention of configuration at the C-2 position when various aluminium hydride were used. The *cis* relationship of C-2 and C-6 substituents of the major isomer was confirmed by the preparation of *trans*-2,6-dimethylpiperidine **13a** (vide infra) and by comparison with the products obtained by reaction of oxazolidine **17**<sup>4</sup> with MeMgI (Scheme 5). Indeed, it is known that such a reaction is stereoselective producing the *cis*-2,6- diastereomer **13b** as the major product<sup>12</sup>.

Scheme 5

By analogy with our earlier results <sup>11</sup>, treatment of 12 with NaBH4 in methanol led to formation of a 85:15 mixture of the *trans* compound 13a and its *cis* isomer 13b. This result has been explained by a mechanism involving prior formation of the iminium ion 16 (Scheme 6). In this case, and in contrast with the mechanism proposed in Scheme 4, the N-phenylethanol chain can adopt a conformation in which the steric hindrance between the aromatic ring and the C-6 substituent is minimized. Consequently, the C-6 methyl pseudo axial conformation 16A was favoured as it minimized steric interactions between the chiral auxiliary and the C-6 methyl group. The stereoelectronically controlled approach of the incoming hydride from the axial direction would lead to formation of the major reaction product 13a. The same considerations involving iminium 15B can explain the formation of 8 in the course of the reduction of 7 with NaBH4.

7. 
$$12a,b$$
  $\xrightarrow{NaBH_4}$   $\xrightarrow{R}$   $\xrightarrow{N}$   $\xrightarrow{N}$ 

Scheme 6

Hydrogenolysis (H<sub>2</sub>, Pd(OH)<sub>2</sub>/C) of **13a** and **13b** afforded (+) **9** and **14** in 75 and 80% yield respectively. Compound **9** was identical with the *trans*-2,6-dimethylpiperidine obtained from lactam **6** (IR, MS, NMR) and exhibited an opposite  $[\alpha]_D$  value. For the *cis*-2,6-derivative **14**, all the spectra data correlated well with those previously reported<sup>8</sup>.

In conclusion we have developed a new strategy for the preparation of both enantiomers of *trans*-2,6-dimethylpiperidine. Some informations concerning the configuration of the intermediates have been obtained, it can explain the results, observed during the alkylation and reduction steps. The application of this methodology to the synthesis of various *trans*-substituted piperidines will be reported in due course.

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#### **EXPERIMENTAL SECTION**

Melting points were determined on a Büchi 530 apparatus and were uncorrected. Boiling points were uncorrected. For a column chromatography, silica gel (Merck silica gel 60F234, 70-230 mesh) was used. <sup>1</sup>H-NMR spectra were recorded on a Bruker ARX 250 spectrometer in CDCl<sub>3</sub> with TMS as internal standard. The data were presented in the format: chemical shift (multiplicity, number of protons, coupling constant). <sup>13</sup>C-NMR were recorded on a Bruker ARX 250 spectrometer in CDCl<sub>3</sub>. Optical rotations were measured on a Perkin-Elmer model 241 polarimeter. Concentrations in g/100 mL of solvent were included following the measured optical rotations. Microanalyses were made by Université P. et M. Curie (Centre Régional de Microanalyse). Mass spectra was made by J. C. Tabet (Université P. et M. Curie) or by F. Libot (Faculté de Pharmacie).

## (-)-(3R,8aS)-8a-methyl-3-phenyl-hexahydro-oxazolo-[3,2-a]-pyridin-5-one 5

5-Oxohexanoic acid (10.3 g, 80 mmol) and *R*-(-)-phenylglycinol **4** (10.9 g, 80 mmol) in toluene (100 mL) were refluxed for a night. The solvent was removed *in vacuo* and the diastereomers were isolated (d.e. 60%). After separation by chromatography on silica gel using ethyl acetate/cyclohexane (90 : 10), the *cis* isomer was obtained in 74% yield and crystallized from Et<sub>2</sub>O; mp 62°C ;  $[\alpha]^{20}_D$  -164.9 (c 1.43, MeOH) : <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$  : 1.4 (s, 3H), 1.5-2.0 (m, 2H), 2.1-2.2 (m, 2H). 2.2-2.6 (m, 2H), 3.9 (dd. 1H, J = 7.8 and 8.8 Hz), 4.5 (dd, 1H, J = 1.0 and 8.8 Hz), 5.3 (dd, 1H and 7.8 Hz), 7.2 (m, 5H) : <sup>13</sup>C NMR (CDCl<sub>3</sub>)  $\delta$  : 16.7, 23.1, 30.2, 34.7 58.0, 69.0, 93.4, 125.0, 126.6, 128.0, 139.6, 168.9. Anal. Calcd for C<sub>1</sub>4H<sub>1</sub>7NO<sub>2</sub> : 72.70 (C) ; 7.41 (H) ; 6.06 (N) : Found 72.68 (C) ; 7.62 (H) ; 5.89 (N).

# $(\cdot)$ -(6R)-1-((1R)-2-hydroxy-1-phenyl-ethyl)-6-methyl-piperidin-2-one 6

At -78°C, titanium (IV) chloride (14 mL, 6 eq) was added to lactam 5 (5 g, 21 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (125 mL). After 15 min of complexation, triethylsilane (20 mL, 6 eq) was added to the solution. The reaction mixture was stirred for 90 min at -78°C and 16 h at rt. The solution was hydrolyzed at -20°C by saturated aqueous NH4F (50 mL) for 3 h. Then, the solution was extracted with CH<sub>2</sub>Cl<sub>2</sub>. The organic layers were washed with saturated aqueous NaHCO<sub>3</sub> (20 mL), saturated aqueous (NH4)<sub>2</sub>SO<sub>4</sub> (20 mL) and dried over MgSO<sub>4</sub>. The solvent was removed *in vacuo* and the diastereomers (d.e. 80%) were obtained in 72% yield.

Benzoyl chloride (2.1 mL, 1.2 eq) and pyridine (2,1 mL) in CH<sub>2</sub>Cl<sub>2</sub> (5 mL) were added to crude diastereomeric alcohols (3.6 g, 10 mmol) at 0°C. After stirring for 15 min, the organic layer was washed with hydrochloric acid 2N (5 mL), with saturated aqueous (NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub> (5 mL) and dried over MgSO<sub>4</sub>. Diastereomeric benzoates were isolated (d.e. 80%). The major one was isolated by crystallization from Et<sub>2</sub>O in 84% yield (mp 101°C).

The latter was saponified with 1% NaOH in MeOH for 1 h. The solvent was removed and the residue was dissolved in H<sub>2</sub>O (5 mL). The aqueous layer was extracted with CH<sub>2</sub>Cl<sub>2</sub>. The combined organic layers were then dried over MgSO<sub>4</sub> and the solvent was removed *in vacuo*. The alcohol was crystallized from Et<sub>2</sub>O in 95% yield (d.e. > 98%); mp 82°C; [ $\alpha$ ]<sup>20</sup>D -12.2 (c 1.3, MeOH); <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$ : 1.1 (d, 3H, J = 6.5 Hz), 1.5-2.0 (m, 4H), 2.3-2.6 (m, 2H), 3.4 (m, 1H), 4.0 (m, 1H), 4.3 (m, 1H), 4.5 (m, 1H), 4.6 (m, 1H), 7.2 (m, 5H); <sup>13</sup>C NMR (CDCl<sub>3</sub>)  $\delta$ : 16.3, 19.8, 29.6, 32.2, 53.4, 64.3, 65.7, 127.5, 128.6, 137.5, 171.8. Anal. Calcd for C<sub>1</sub>4H<sub>1</sub>9NO<sub>2</sub>: 72.07 (C); 8.21 (H); 6.00 (N); Found 71.98 (C); 8.19 (H); 5.93 (N).

# (-)-(3R,5R,8aS)-(5,8a)-dimethyl-3-phenyl-hexahydro-oxazolo[3,2-a]-pyridine 7

Sodium hydride (171 mg, 4.3 mmol) (60% by weight in mineral oil) was added to alcohol **6** (1 g, 4.3 mmol) in THF (10 mL). After stirring at rt for 1 h, MeMgCl (3M/THF) (10 mL, 7 eq) was added to the mixture and the solution was refluxed for 12 h. After cooling, the reaction mixture was quenched with saturated aqueous NH4Cl (20 mL) and with hydrochloric acid 2.4 N (10 mL). The solution was extracted with CH<sub>2</sub>Cl<sub>2</sub>. The combined organic layers were dried over MgSO<sub>4</sub>, the solvent removed *in vacuo* and the residue chromatographed on silica gel using ethyl acetate/cyclohexane (90 : 10). Compound 7 was crystallized from pentane in 55% yield (d.e. > 98%); mp 47°C :  $[\alpha]^{20}D$  -101.5 (c 1.O4, MeOH); <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$  : 0.6 (d, 3H, J = 6.9 Hz), 1.1-1.8 (m, 6H), 1.4 (s,3H), 3.1 (m, 1H), 3.5 (dd, 1H, J = 9.5 and 12.4 Hz), 4.2 (dd, 2H, J = 3.8 and 12.4 Hz) 7.1-7.4 (m, 5H). <sup>13</sup>C NMR (CDCl<sub>3</sub>)  $\delta$  : 22.1, 22.3, 24.1, 26.3, 29.7, 51.4, 59.4, 72.7, 95.5, 126.7, 126.9, 128.2, 147.0. Anal. Calcd for C<sub>1</sub>5H<sub>2</sub>1NO: 77.88 (C); 9.15 (H); 6.05 (N); Found 77.53 (C); 9.05 (H); 5.78 (N).

# (-)-(2'R)-((2R,6R)-2,6-dimethyl-piperidin-1-yl)-2'-phenyl-ethanol 8

A: Oxazolopiperidine 7 (400 mg, 1.7 mmol) in Et<sub>2</sub>O (10 mL) was slowly added at 5°C to LiAlH<sub>4</sub> (100 mg, 2.6 mmol) in suspension in Et<sub>2</sub>O (10 mL). The mixture was stirred at this temperature for 1 h. The reaction mixture was then quenched with H<sub>2</sub>O (5 mL) and with NaOH 3N (5 mL). After filtration through celite, the solution was extracted with Et<sub>2</sub>O. The organic layer was dried over MgSO<sub>4</sub> and the solvent removed *in vacuo*. Compound 8 was isolated in 82% yield as a colorless oil (d.e. > 98%). [ $\alpha$ ]<sup>20</sup>D -79.5 (c 1.27, MeOH); <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$ : 0.7 (d, 6H, J = 6.4 Hz), 1.1-1.3 (m, 2H), 1.4-1.7 (m, 4H), 3.2 (m, 2H), 3.5 (dd, 1H, J = 5.0 and 9.9 Hz), 3.9 (t, 1H, J = 9.9 Hz), 4.3 (dd, 1H, J = 5.0 and 9.9 Hz), 7.1-7.3 (m, 5H); <sup>13</sup>C NMR (CDCl<sub>3</sub>)  $\delta$ : 17.5, 19.2, 35.9, 46.1, 58.3, 58.4, 127.4, 127.9, 128.0, 128.5, 140.3. Anal. Calcd for C<sub>1</sub>5H<sub>23</sub>NO: 77.20 (C): 9.94 (H): 6.00 (N); Found 77.39 (C): 10.14 (H): 5.73 (N).

B: NaBH<sub>4</sub> (114 mg, 3 mmol) was added to a cooled (5°C) solution of oxazolidine 7 (100 mg, 0.4 mmol) in MeOH (30 mL). The resulting mixture was stirred at this temperature for 3 h then quenched with  $H_2O$  (10 mL) and the solvent removed *in vacuo*. The aqueous layer was then dried over MgSO<sub>4</sub> and the solvent removed *in vacuo*. Compound 8 was isolated in 80% yield (de > 98%).

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#### (-)-(2R,6R)-2,6-dimethylpiperidine or lupetidine 9

Alcohol **8** (850 mg, 3.6 mmol) in ethanol (5 mL) was added to palladium on charcoal (10%) (850 mg, 1 eq) in suspension in ethanol (20 mL). Concentrated hydrochloric acid (5 mL) was added and the mixture was hydrogenolyzed at rt under atmospheric pressure for 12 h. After filtration on celite, the solvent was removed *in vacuo*. The residue was dissolved in H<sub>2</sub>O and the aqueous layer was washed with Et<sub>2</sub>O. Then, the acid layer was neutralized with solid K<sub>2</sub>CO<sub>3</sub> and the reaction mixture extracted with Et<sub>2</sub>O. The combined organic layers were dried over MgSO<sub>4</sub> and the solvent distilled under atmospheric pressure. Compound **9** was purified by distillation and obtained in 60% yield; bp 74°C (85 mm Hg). [ $\alpha$ ]<sup>20</sup>D -12 (c 0.5, EtOH) (lit<sup>11</sup> -14° (EtOH)); <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$ : 1.0 (d, 6H, J = 6.5 Hz), 1.2 (m, 2H), 1.5 (m, 4H), 3.1 (m, 2H); <sup>13</sup>C NMR (CDCl<sub>3</sub>)  $\delta$ : 19.4, 21.0, 32.8, 46.0. Anal. Calcd for C7H<sub>1</sub>5N: 74.27 (C); 13.36 (H); 12.37 (N); Found 74.04 (C); 13.31 (H); 12.23 (N); MS m/e 114.98.

# (3R,5S)-(5,8a)-dimethyl-3-phenyl-hexahydro-oxazolo[3,2-a]-pyridine 12a,b

Sodium hydride (238 mg, 6.18 mmol) (60% by weight in mineral oil) was added at 0°C under a nitrogen atmosphere to a solution of lactame 11 (1.2 g, 5.15 mmol) in THF (100 mL). After stirring at rt for 1 h, MeMgI (17 mL, 10 eq, 3M/Et<sub>2</sub>O) was carefully added, giving a white precipitate. This mixture was vigorously stirred for 2 h. After cooling, the resulting suspension was quenched with saturated aqueous NH<sub>4</sub>Cl (40 mL) then extracted twice with EtOAc. The combined organic layers were dried over MgSO<sub>4</sub> and the solvent removed *in vacuo*. The oily residue was purified by flash-chromatography on silica gel using EtOAc to furnish oxazolidine 12 (536 mg, 2.32 mmol, 45%) as a 80:20 mixture of unseparable diastereomers, and recovered lactam 11 (443 mg, 1.9 mmol, 37%). <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$ : 0.55 and 0.95 (2d, J = 6 Hz, 6H, Me), 0.85 (m, 2H, H-4), 1.5-1.9 (2m, 4H, H-3, H-5), 2.80 (m, 1H, H-2). 3.64 (t, J = 6.9 Hz, 1H, H-8), 4.18 (dd, J = 7.7, 6.9 Hz, 1H, H-8), 4.23 (dd, J = 7.7, 6.9 Hz, 1H, H-7), 7.15-7.40 (m, 5H, Ar). <sup>13</sup>C NMR (CDCl<sub>3</sub>)  $\delta$ : 17.9, 22.7, 28.0, 29.2, 33.1, 53.1, 67.9, 70.4, 95.0, 126.8-128.3, 143.8. MS m/z (relative intensity), 231(M+•), 216(70), 188(35), 162(45), 146(50).

## (-)-(2'R)-((2S,6S)-2,6-dimethyl-piperidin-1-yl)-2'-phenyl-ethanol 13a

To a cooled solution of the oxazolidine mixture **12a,b** (210 mg), 0.91 mmol) in MeOH (15 mL) an excess of NaBH4 (200 mg) was added at 0°C. The resulting mixture was stirred at rt for 2 h then quenched with saturated aqueous NH4Cl (10 mL) and extracted with CH<sub>2</sub>Cl<sub>2</sub>. The combined organic layers were dried over MgSO<sub>4</sub> and the solvent removed *in vacuo*. The residue was purified by column chromatography (Al<sub>2</sub>O<sub>3</sub>, CH<sub>2</sub>Cl<sub>2</sub>:cyclohexane, 50:50) to give **13** (150 mg, 72%) as a 85:15 mixture of diastereomers.

The *trans* diastereomer **13a** was isolated pure after preparative tlc (Al<sub>2</sub>O<sub>3</sub>) using the same eluant; colorless oil,  $[\alpha]_D$ : -19 (c = 1.35, CHCl<sub>3</sub>). <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$ : 1.0-1.4 (3m, 6H, H-3, H-4, H-5), 1.20 (d, J = 7 Hz, 6H, CH<sub>3</sub>), 3.32 (m, 2H, H-2, H-6), 3.52 (dd, J = 9.9, 5.6 Hz, 1H, H-8), 3.85 (t, J = 9.9 Hz, 1H, H-8), 4.20 (dd, J = 9.9, 5.6 Hz, 1H, H-7), 7.15-7.40 (m, 5H, Ar). <sup>13</sup>C NMR (CDCl<sub>3</sub>)  $\delta$ : 20.0, 30.2, 47.8, 59.6, 60.3, 127.2-129.0, 141.3. MS, m/z (relative intensity): 234(M+1, 100), 202(25). Anal. Calcd for C<sub>15</sub>H<sub>23</sub>NO: 77.21 (C): 9.93 (H): 6.00 (N): Found 76.95 (C): 9.81 (H): 6.01 (N).

## (-)-(2'R)-((2R,6S)-2,6-dimethyl-piperidin-1-yl)-2'-phenyl-ethanol 13b

A: LiAlH<sub>4</sub> (150 mg, 5.70 mmol) was proportionwise added to a cooled (-5°C) solution of oxazolidine **12a,b** (205 mg, 0.88 mmol) in THF (30 mL) under an argon atmosphere. The mixture was stirred at rt for 2 h and then treated successively with H<sub>2</sub>O (0.15 mL), 15% NaOH (0.15 mL) and H<sub>2</sub>O (0.45 mL). After filtration on a short pad of MgSO<sub>4</sub>, the residual white solid was washed several times with Et<sub>2</sub>O. Removal of the solvent *in vacuo* gave a product which was purified by column chromatography (Al<sub>2</sub>O<sub>3</sub>, CH<sub>2</sub>Cl<sub>2</sub>:cyclohexane, 50:50) to yield 135 mg of **13** (0.58 mmol, 68%) as a 20:80 mixture of diastereomers.

**B**: MeMgI in Et<sub>2</sub>O (1.5 mL, 4.5 mmol) was slowly added to a cooled (-5°C) solution of oxazolidine **17** (260 mg, 1.19 mmol) in THF (20 mL) under an argon atmosphere. After stirring at rt for 2 h, the mixture was quenched with saturated aqueous NH<sub>4</sub>Cl (10 mL) and extracted with CH<sub>2</sub>Cl<sub>2</sub>. The combined organic layers were dried over MgSO<sub>4</sub> and the solvent removed *in vacuo*. The oily residue was filtred on a short pad of alumina (CH<sub>2</sub>Cl<sub>2</sub>) to give a 75:25 mixture of *cis* **13b** and *trans* **13a** diastereomers (220 mg, 78%). The major one **13b** was isolated pure after preparative tlc (Al<sub>2</sub>O<sub>3</sub>, CH<sub>2</sub>Cl<sub>2</sub>:cyclohexane, 50:50); colorless oil, [α]<sub>D</sub>: -24 (c = 1.45, CHCl<sub>3</sub>). H NMR (CDCl<sub>3</sub>) δ : 1.08 and 1.18 (2d, J = 7 Hz, 6H, Me-2, Me-6), 1.30-1.45 (m, 4H, H-4, H-5), 1.65 (m, 2H, H-3), 2.88 (m, 1H, H-6), 3.05 (m, 1H, H-2), 3.80 (dd, J = 8.6, 6.3 Hz, 1H, H-8), 3.86 (t, J = 8.6 Hz, 1H, H-8), 4.22 (dd, J = 8.6, 6.3 Hz, 1H, H-7), 7.20-7.35 (m, 5H, Ar). H NMR (CDCl<sub>3</sub>) δ : 19.2, 21.1, 32.9, 33.7, 50.2, 53.0, 61.8, 62.4, 127.3-129.0, 139.8. MS, m/z (relative intensity) : 234 (M<sup>+1</sup>, 90), 202(30). Anal. Calcd for C<sub>1</sub>5H<sub>2</sub>3NO : 77.21 (C) ; 9.93 (H) ; 6.00 (N) ; Found 77.14 (C) : 9.78 (H) ; 5.82 (N).

## References

- 1) L. Micouin, J.-C. Quirion, H.-P. Husson, Tetrahedron Lett., 1996, 37, 849-852.
- a) M.J. Munchhof, A.I. Meyers, J. Org. Chem., 1995, 60, 7084-7085;
  b) L. E. Burgess, A. I. Meyers, J. Org. Chem., 1992, 57, 1656-1662.
- 3) S. Fréville, J.-P. Célérier, V.M. Thuy, G. Lhommet, *Tetrahedron: Asymmetry*, 1995, 6, 2651-2654.

- 4) P. Lienard, T. Varea, J.-C. Quirion, H.-P. Husson, Synlett, 1994, 143-144.
- 5) M. Bonin, D.-S. Grierson, J. Royer, H.-P. Husson, Org. Synth., 1991, 70, 54.
- 6) G. Lunn, E.B. Sansone, J. Org. Chem., 1986, 51, 513-517.
- 7) C. Boga, F. Manescalchi, D. Savoia, Tetrahedron, 1994, 50, 4709-4722.
- 8) B. Guilloteau-Bertin, PhD Thesis, Université Paris-XI Orsay, 1995.
- a) S. Najdi, M.J. Kurth, Tetrahedron Lett., 1990, 31, 3279-3282;
  b) R. K. Hill, T. Yuri, Tetrahedron. 1977, 33, 1569-1571.
- 10) A.D. Kuzovkov, G.P. Men'shikov, J. Gen. Chem., 1950, 20, 1524-1527.
- 11) D.S. Grierson, J. Royer, L. Guerrier, H.-P. Husson, J. Org. Chem., 1986, 51, 4475.
- 12) L. Guerrier, J. Royer, D.S. Grierson, H.-P. Husson, J. Am. Chem. Soc., 1983, 105, 7754-7755.

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